

REIDER, L.; DOMITROV, J.

Effect of denervation and immobilization on carbohydrate metabolism in tonic and tetanic muscles. Part 1 and 2. Acta physiol. scand. 88 no.3:227-244 '65.

1. Department of Neurology and Brain Research, University Medical School, Sweden. Submitted January 8, 1965.

HEINER, L.; STIPULA, Magda; DOMONKOS, J.

Connection between carbohydrate metabolism and atrophy in tonic and tetanic muscles. *Acta physiol. acad. sci. Hung.* 28 no.3:245- 252 : 65.

1. Department of Neurology and Brain Research Institute, University Medical School, Szeged, and Department of Neurology, University Medical School, Budapest.

LATZKOVITS, L.; DOMONIKOS, J.

The effect of postnatal development on the carbohydrate metabolism of tinnic and tetanic muscles. Acta physiol. scand. sci. 119g. 28 no. 3:253-257 ' 65.

1. Department of Neurology and Brain Research Institute
University Medical School, Ssaged.

Submitted January 8, 1965/

L 31080-66

ACC NR: AT6022818

SOURCE CODE: HU/2505/65/028/003/0227/0236

AUTHOR: Domonkos, Jeno--Domonkosh, Yo.; Heiner, Lajos--Kheyner, L. 25

ORG: Neurological Clinic, Medical University, Szeged (Orvostudományi Egyetem
Idősklinikája); Brain Research Institute, Medical University, Szeged (Orvostudományi
Egyetem Agukutató Intézet) B+1

TITLE: Effect of denervation and immobilization on carbohydrate metabolism in tonic and tetanic muscles I. Glycolytic metabolism 22

SOURCE: Academia scientiarum hungaricae. Acta physiologica, v. 28, no. 3, 1965, 227-236

TOPIC TAGS: carbohydrate, muscle physiology, biologic metabolism

ABSTRACT: Denervation does not influence to the same extent the glycogen content and glycolytic metabolism of tonic and tetanic muscles. During the first week following denervation, the glycogen content decreases in the tonic muscle and increases in the tetanic one. In the second week after denervation, the situation is reversed. The glycolytic metabolic activity and the changes in glycogen content show a parallel behavior in both kinds of muscle. the second phase after denervation, the glycolytic metabolism of the two kinds of muscle begins to show similarities. With a few days' lag, the effect of immobilization resembles that of denervation. Orig. art. has: 8 tables. [Orig. art. in Eng.] [JPRS]

SUB CODE: 06 / SJEM DATE: 08Jan65 / ORIG REF: 003 / SOV REF: 001

OTH REF: 025

Card 1/1 CC

L 31081-66

ACC NR: AT6022819

SOURCE CODE: HU/2505/65/028/003/0237/0244

AUTHOR: Heinor, La[os]—Kheyner, L.; Domonkos, Jeno—Domonkosh, Ye. 24
B+1

ORG: Department of Neurology and Brain Research Institute, Medical University of Szeged (Szegedi Orvostudományi Egyetem, Idegklinika és Agykutatási Intézet) 22

TITLE: Effect of denervation and immobilization on the carbohydrate metabolism in tonic and tetanic muscles II. Oxidative metabolism

SOURCE: Academia scientiarum hungaricae. Acta physiologica, v. 28, no. 3, 1965, 237-244

TOPIC TAGS: muscle physiology, carbohydrate, biologic metabolism

ABSTRACT: In the second postoperative week following denervation, there is a significant difference in the oxygen uptake by tetanic and tonic muscles; oxygen uptake decreases in the tonic and slightly increases in the tetanic muscle. If inactivity is enforced by immobilization, however, there are no changes in oxygen uptake by either kinds of muscle as much as two weeks after fixation. The high energy phosphate content of tonic and tetanic muscles shows no changes in opposite directions. As a result of inactivation, a slight increase in high energy phosphate content can be noted in both kinds of muscle. This increase is more marked in the tonic muscle. Orig. art. has: 4 tables. [Orig. art. in Eng.] [JPRS]

SUB CODE: 06 / SUFEM DATE: 08Jan65 / ORIG REF: 003 / SOV REF: 003
OTH REF: 026

L 31082-66

ACC NR: AT6022820

SOURCE CODE: HU/2505/65/028/003/0245/0252

AUTHOR: Heiner, Lajos--Kheyner, L.; Stipula, Magda--Shtipula, M.; Domonkos, János--Domonkosh, Y. 25

ORG: Heiner, Domonkos Neurological Clinic, Medical University, Szeged (Orvostudományi Egyetem Idegklinikája); Heiner, Domonkos Brain Research Institute, Medical University, Szeged (Orvostudományi Egyetem Agykutatási Intézete); Stipula Neurological Clinic, Medical University, Budapest (Orvostudományi Egyetem Idegklinikája) 8+1

TITLE: Correlation between carbohydrate metabolism and atrophy in tonic and tetanic muscles 22

SOURCE: Academia scientiarum hungaricae. Acta physiologica, v. 28, no. 3, 1965, 245-252

TOPIC TAGS: biologic metabolism, muscle physiology, carbohydrate

ABSTRACT: The relationship between atrophy, metabolism, and other changes brought about by denervation have been examined in tonic and tetanic muscles. Atrophy was found to appear earlier in the tonic muscle than in the tetanic one. During the 2-3 weeks' observation period, the atrophic process did not in all respects follow the changes which occurred in metabolism. Changes in the glycolytic metabolism appear earlier after denervation than does a histologically demonstrable atrophy. Changes in oxidative metabolism appear after the process

Card 1/2

L 31082-66

ACC NR: AT6022820

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of atrophy has become histologically demonstrable. The changes in glycolysis and oxygen uptake, following denervation, are inverse in the two kinds of muscle. These changes can not, therefore, be directly related to atrophy. On the other hand, a closer correlation may exist between the high energy phosphate content, the muscle cell phase, and atrophy. Orig. art. has: 13 figures and 1 table. [Orig. art. in Eng.] [JPMS]

SUB CODE: 06 / SUHM DATE: 08Jan65 / ORIG REF: 002 / OTH REF: 007

Card 2/2 CC

L 31083-66

ACC NR: AT6022821

SOURCE CODE: HU/2505/65/028/003/0253/0257

AUTHOR: Latzkovits, Laszlo--Latskovich, L.; Domonkos, Jeno--Domonkosh, Y. ²⁵
^{B+1}

ORG: Neurological Clinic, Medical University, Szeged (Orvostudományi Egyetem Ideg-
Klinikája); Brain Research Institute, Medical University, Szeged (Orvostudományi
Egyetem Agykutató Intézet)

TITLE: Effect of postnatal development on the carbohydrate metabolism of tonic
 and tetanic muscles ²²

SOURCE: Academia scientiarum hungaricae. Acta physiologica, v. 28, no. 3, 1965, 253-257

TOPIC TAGS: carbohydrate, biologic metabolism, muscle physiology

ABSTRACT: Lactate and pyruvate production under aerobic conditions is the most characteristic metabolic process of the so-called tetanic muscles of the skeletal musculature. No such process can be observed in tonic muscles. On examining the metabolism of the skeletal musculature on the basis of its lactate and pyruvate production during postnatal development, it has been established that, immediately after birth, the metabolism of the skeletal muscles corresponds entirely to that of the tonic muscles. The characteristic metabolism of the tetanic muscle is the result of postnatal differentiation. This differentiation of metabolism occurs parallel with the differentiation of muscle function. Previous to the development of independent motion and locomotor function, tetanic and tonic muscles can not be dealt with separately. Orig. art. has: 2 tables. Orig. art. in Eng. / JPRS
 SUB CODE: 06 / SUM DATE: 08 Jan 65 / ORIG REV: 004 / 30V REF: 001 / OTH REF: 011

ACC NR: AP6030837

SOURCE CODE: RU/0023/66/011/001/0041/0046

AUTHOR: Sabau, Monica--Sabeu, M. (Doctor); Domokos, L.--Domokosh, L. (Doctor);
Abraham, A.--Abragam, A. (Doctor); Nagy, L.--Nad^l, L. (Doctor)

ORG: IME, Tirgu Mures

TITLE: Etiological role of atypical Esch. coli (of hemolytic type) in pediatric enterocolitis. [This paper was presented at a meeting at Section of Infectious Pathology, U.S.S.M., Mures-Autonomia Branch, Hungary, on 8 October 1964.]

SOURCE: Microbiologia, parazitologia si epidemiologia, v. 11, no. 1, 1966, 41-46

TOPIC TAGS: pediatrics, gastroenterology, blood chemistry, bacteria

ABSTRACT: In a study of enterocolitis in children, hemolyzing strains of Esch. coli were isolated in 31.0% percent of 960 cases. A relatively larger amount of hemolysis was produced in young cultures. The authors suggest that the hemolysin isolated from supernatant cultures in alkaline broth is of type α and probably a protein. The hemolysin was found to be thermolabile. Orig. art. has: 3 tables. [Based on authors' Eng. abst.] [JPRS: 35,814]

SUB CODE: 06 / SUBM DATE: 26Oct64 / ORIG REF: 007 / SOV REF: 002
 OTH REF: 010

Cord 1/1 RS

UDC: 616.348-002-02:576.851.48

DOMOKOSH, G.

Calculation of Regge poles. Dokl. AN SSSR 145 no.1:76-77 J1
'62. (MIRA 15:7)

1. Obyedinennyy institut yadernykh issledovaniy. Predstavleno
akademikom N.N. Bogolyubovym.
(Diffraction) (Quantum theory)

DOMOKOSH, G.

Analytic properties of the amplitude of elastic $\pi\pi$ -scattering in a 1-plane. Dokl. AN SSSR. 144 no.6: 1279-1280 Je '62. (MIRA 15:6)

1. Ob"yedinennyy institut yadernykh issledovaniy. Predstavleno akad. N.N.Bogolyubovym.

(Mesons--Scattering)

DOMOKOSI, G. [Domokos, G.]

On the asymptotic behavior of the vertex part in the pseudoscalar meson theory. Acta phys Hung 14 no.4:341-344 '62.

1. Ob"yedinennyy institut yadernykh issledovaniy, Laboratoriya teoreticheskoy fiziki, Dubna, SSSR. Komandirovan iz Tsentralnogo nauchno-issledovatel'skogo instituta fiziki Vengerskoy Akademii Nauk Predstavleno L. Yanoshi [Lajos Janossy].

DOMONKOS, Erica

Mathematics methods in the realization of intermediate language for
automatic translation. Probleme automatiz 4:63-68 '63.

MASSZI, Jozsef, dr.; DOMONKOS, Robert, dr.

Epithelial changes following periorbital dermatitis in old age.
Borogygy. vener. szemle 38 no.3:123-124, J1 '62.

1. A Budapesti Orvostudományi Egyetem Bor- és Nemikortani Klinikájának
(Igazgató: Foldvari Ferenc dr. egyetemi tanár) közleménye.
(DERMATITIS in aged) (ORBIT dis)

LENGYEL, Julia, dr.; KALDOR, Istvan, dr.; DOMONKOS, Robert, dr.

Simultaneous occurrences of actinomycosis and blastomycosis.
Orv. hetil. 104 no.44:2091-2092 3 N '63.

1. Budapesti Orvostudományi Egyetem, Bor- és Nemikortani Klinika.
 (ACTINOMYCOSIS) (BLASTOMYCOSIS)
 (APPENDECTOMY) (PATHOLOGY) (PENICILLIN)
 (STREPTOMYCIN) (ACTINOMYCIN)

DOMONKOS, Robert, dr.

Rare localization of deep trichophytosis. Borgyogy, vener.
szemle 40 no.2:86-88 Ap'64

1. Budapesti Orvostudományi Egyetem Bor- és Nemikortani Kli-
nikájának (Igazgató: dr. Foldvari, Ferenc, egyetemi tanár) .
közleménye.

*

DOMONKOS, Robert, dr.; BOTTYAN, Etelka, dr.

Drug side-effects in 2 cases of subacute lupus erythematosus.
Borgyogy vener. szemle. 40 no.4:180-182 Ag '64.

1. A Budapesti Orvostudományi Egyetem Bor- és Nemikortani Klinika
(Igazgató: Foldvári Ferenc dr. egyetemi tanár) közleménye.

DOMONKOS, Sandor

Recurrent voltage originating during the break of testing circuits
by means of a contactor. Elektrotechnika 54 no.12:575-576 D '61.

UJHASI, Geza; DOMONKOS, Sandor

Review of periodicals. Elektrotechnika 54 no. 12:574-576 D'61.

DOMONKOS, Sandor

"Thermo-bimetal in electrical engineering" by Dr. Frantisek
Kaspar. Reviewed by Sandor Domonkos. Elektrotechnika 54
no. 6:287 Jo '61.

DOMONKOS, S. (Budapest XI., Egry Jozsef u.18)

The effect of magnetic blowout on the arc extinction time in alternating-current protective switches. Periodica polytechn electr 6 no.2:125-148 '62.

1. Lehrstuhl für Hochspannungstechnik und Geräte, Technische Universität, Budapest Vorgelegt von Prof. DR.J.Eisler.

DOMONKOS, Sandor, adjunktus

Quenching low-voltage arcs. Elektrotechnika 53 no.11:503-513 '60.

1. Budapesti Műszaki Egyetem Nagyfeszültségű Technika és Készülékek Gyára.

DOMONKOS, Sándor (Budapest, XT., Egry József u.18)

Forces occurring between the deion quenching plates and the light arc. Periodica polytechnica electr 8 no.1:79-92 '64.

1. Lehrstuhl für Hochspannungstechnik und Elektrische Apparate, Universität Budapest. Vorgelegt von Prof. Dr. J. Eisler.

ANTAL, Ferenc (Biborteni 175. Rai:Sft. Gheorghe. Reg. Aut. Magh.-Rumania);
SZEMLER, Ferenc; DOMONKOS, Zoltan (Szigetszentmiklos, Jozsef A.
telep); LIPPERT, Jozsef (Budapest V., Balaton u.27); HORVATHY,
Lajos (Budapest XX., Marx Karoly u.180)

Motorists' letters. Auto motor 14 no.4:5 F '61.

DOMOKOS, Zoltan, mr

Packaging of pharmaceutical products slated for export.
Farmaceut gl Zagreb 20 no.9:335-337 S '64.

1. "Galenika" Enterprise, Belgrade.

DOMONSKIY, E. I.

Category : USSR/Optics - Physical optics

K-5

Abs Jour : Ref Zhur - Fizika, No 1, 1957, No 2304

Author : Domonskiy, E.I., Neskov, M.M.

Inst : Ural' State University, USSR

Title : Determination of the Optical Constants of Metals, Using the Autocollimation Method

Orig Pub : Fiz. metallov i metallovedeniye, 1955, 1, No 3, 567

Abstract : The autocollimation method (O'Bryan H.M., J. Opt. Soc. America, 1936, 26, 122) is used in the infrared region. Light is reflected from the specimen twice in the forward path and in the return path, and passes twice through the same Se polarizer, which is slowly rotated by an electric motor. A recording infrared spectrometer automatically records the intensity, the minimum deviation of which is used to find the principal azimuth ψ_0 and the principal angle of incidence ϕ_0 with a probable error of $\pm 35'$ for ψ_0 and $\pm 15'$ for ϕ_0 in each individual measurement. The optical constants of mirrors obtained by evaporation of Sb, Cu, Ag, Al, and Zn in vacuum were measured for λ 2.45 μ .

Card : 1/1

DOMONTOVICH, Ye. N. Doc Med Sci -- (diss) "Data ⁷⁰Concerning the
Proble m of the Adaptation of the Organism to the Hypoxic Form
of Oxygen Deficiency." Mos, 1957. 21 pp 20 cm. (Academy of
Medical Sciences USSR), 200 copies (KL, 18-57, 97)

- 47 -

DOMONTOVICH, Ye., doktor med.nauk; PAKHOVA, L.N., kand. biol. nauk

Characteristics of physiological functions of daily periodicity
in patients recovered from meningitis. Vrach delo, no.7:79-83
Jl'63. (MIRA 16:10)

1. Fiziologicheskoye otdeleniye (zav. - doctor med. nauk Ye.K.
Domontovich) Tsentral'nogo nauchno-issledovatel'skogo institu-
ta ekspertizy trudesposobnosti i organizatsii truda invalidov.
(PHYSIOLOGY, PATHOLOGICAL) (MENINGITIS)

1st AND 2ND EDITION		PROCESSING AND PROPERTIES INDEX	
<i>Correction of Aluminium and Its Prevention.</i> André Dumont (Hans). <i>Kolloid. Zepf.</i> 1937, 70, 319-324; C. Ab., 1938, 88, 2073. The electrolytic theory of the corrosion of aluminium is described. A 20 cm.² aluminium surface developed in sea-water within 6 months about 25 c.c. gaseous hydrogen. This gas exerted a pressure in the metal sufficient to cause protuberances on the surface. Impurities (aluminium ferrite crystals formed during slow cooling, rolled-in copper particles, etc.) promote corrosion. Local rubbing, improper storage, etc., can develop local heat sufficient to cause oxidation, converting the metal to a graphite-like powder and producing black spots that sometimes extend some thousandths of a mm. into the metal. The use of iron or copper rivets in aluminium plates causes corrosion. Metal rolled at 500° C. is much more resistant than that worked at 300° C. In alloys designed for corrosion-resistance (manganese-magnesium-aluminium, manganese-aluminium, magnesium-aluminium) intercrystalline corrosion due to improper heat-treatment may occur. Baked lacquers from phthalic acid, phenol-formaldehyde, or chlororubber resins may prevent corrosion, but the heat in baking adversely affects the mechanical properties of the metal. Aluminium plated with nickel, cadmium, or zinc corrodes very quickly if the slightest hole is formed in the plating. Surface oxidation by means of chemical agents or by electrolysis seems at present to be the best method of corrosion prevention.—S. G.		41	
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION		RESEARCH	
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1ST AND 2ND ORDERS		PROCESSES AND PROPERTIES INDEX	
Aluminum 99.99% pure. Andras Jomony. <i>Bany-ds, Kohls. Lapok 71, 281-4(1938)</i>. The characteristics of pure Al made in the Csepel factory are described, with photomicrographs of its structure. In regard to alloys of pure Al, Fe improves its mech. qualities but decreases its chem. resistance. Addn. of 0.1% Fe, e. g., increases the HCl soly. 500-fold. The breaking strength of 99.99% Al is 12.0; that of Al contg. 0.10% Fe, 14.5; and that of Al contg. 0.005% Fe, 17.5. Alloying with Cu similarly decreases chem. resistance and increases strength (0.10%, 14.4; 0.50, 17.0). Addn. of Si treated at 300° greatly decreases the chem. resistance, but Si treated at 300° increases strength (0.20% Si, 15.5; 0.005% Si, 18.6). Mg increases the strength, 0.10% Mg, 14.0; 0.47% Mg, 17.4). Al of 99.99% purity seems to be practically suitable for covering surfaces and profiles and is resistant against atm. influences. Thin Al foils of pure metal are good for covers for foods. S. S. de Finais		9	
ASS-SLA METALLURGICAL LITERATURE CLASSIFICATION			
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101000 000 000 000		0111111 000 000 000	

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***Blister Formation in Semi-Polished Parts of Aluminium or Aluminium Alloys.** Andrei Dumany (*Moskovskii Izvestii*, 1941, 74, 455-460; *Chem. Zentr.*, 1942, 118, (11), 463; *C. Ab.*, 1943, 87, 6231). -- It is shown by experiment that the cause of surface blisters, apart from such well known factors as insufficient deoxidation or faulty casting and fabricating procedures, can be aggregation in the casting. To obtain a sound structure, the castings should be cooled quickly and uniformly.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

1ST AND 2ND CODES		PROCESSING AND PROPERTIES INDEX		1ST AND 2ND CODES	
<i>M</i> <i>2</i> The [Aluminum] Alloy FKG II and its Manufacture. <i>And-As Document</i> <i>(Figs. 1-4 to Krasnodar, 1941, 2, (2), 3-4; Chron. Zentr., 1943, 116, (1), 1978; C. Aba., 1944, 98, 4531).--The alloy FKG II is easily workable, has a</i> <i>silvery lustre, and looks, when polished, like a metal with chromed surface.</i> <i>The strength of sand castings is 15-18 kg./sq. mm. and elongation 1.5-2.5%;</i> <i>when cast in metal moulds the corresponding figures are 18-20 kg./sq. mm.</i> <i>and 1.5-3%. Modulus No. -- 7-8, density -- 2.75, shrinkage -- 3%. During</i> <i>melting, casting, and working it behaves like other aluminium alloys. The</i> <i>correct melting and casting temperatures are 720-730° C. and 710-720° C.</i> <i>respectively.</i>					
ASH 51.4 METALLURGICAL LITERATURE CLASSIFICATION					

CA

9

Recovery and refining of Al wastes and cuttings.
 Andras Thony, *Technica* (Budapest) 23, 420-41
 (1962) *Chem. Zvest.* 1962, 11, 1780. The Al wastes
 must be sep'd. into groups according to their compn. in
 alloys: pure Al, Al-Cu to 10% Cu, Al-Cu-Mg, Al-Si-Mg
 (Cu-free), Al-Mg (Cu-free with high Mg content), Al-Si
 (high Si content) Al-Ni-Mg and alloys of unknown compn.
 Two tech. processes were worked out—the fractional
 crystn. and the refining by means of a 3-layer electrolysis.
 However, none of the processes gives a satisfactory solu-
 tion to the problem.

S. Pakawer

ASD 56.6 METALLURGICAL LITERATURE CLASSIFICATION

FROM SOURCE

FROM SOURCE

ALLOY

ALLOY

ALLOY

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9

Corrosion of light metals and its prevention. Anlauf, Hermann. *Angew. Chem. Leipzig* 76, 221-30 (1964). *Chem. Zentr.* 1964, II, 1446. - Various types of corrosion (chem., electrochem. and intercryst.) occurring in light-metal alloys and the effect of alloying elements, impurities and heat-treatment are discussed. Protective methods are described in detail and operating suggestions are given for anodic oxidation, chem. oxidation, plating, galvanizing and lacquering.
H. W. Rathmann

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

157 AND 158 DATES

PROCESSES AND PROPERTIES INDEX

9

Working up and refining of light-metal wastes. András Ilanony and Elemér Kovács. *Aluminiun Készítés* 1, IV-4, 1-16(1947); cf. C.A. 30, 6760. Detailed procedure is described. The importance of a previous exact classifying of wastes of various compn. is emphasized. Av. yields are: 77-82% in crucible furnaces of small casting plants, 85-91% in trough furnaces of medium-sized works, and 90-95% in rotating furnaces of large plants. István Flukly

COMMON ELEMENTS

NON-FERROUS METALS

FERROUS METALS

STEEL

IRON

CO

CH

AV

NO

AL

SI

TI

TA

W

MO

Cr

MN

Fe

Co

Ni

Cu

Zn

Al

Mg

Ca

Na

K

Li

B

P

S

Se

Te

As

Sb

Bi

Pb

Sn

Hg

Ag

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Pt

Ir

Rh

Os

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Mo

W

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Au

Pt

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W

Cr

Mn

Fe

Co

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Li

B

P

S

Se

Te

As

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Pb

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Ag

Au

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Cr

Mn

Fe

Co

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Na

K

Li

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P

S

Se

Te

As

Sb

Bi

Pb

Sn

Hg

Ag

Au

Pt

Ir

Rh

Os

Re

Mo

W

Cr

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Zn

Al

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Na

K

Li

B

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Se

Te

As

Sb

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Mo

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Na

K

Li

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Ir

Rh

Os

Re

Mo

W

Cr

Mn

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Cu

Zn

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Mg

Ca

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K

Li

B

P

S

Se

Te

As

Sb

Bi

Pb

Sn

Hg

Ag

Au

Pt

Ir

Rh

Os

Re

Mo

W

Cr

Mn

Fe

Co

Ni

Cu

Zn

Al

Mg

Ca

Na

K

Li

B

P

S

Se

Te

As

Sb

Bi

Pb

Sn

Hg

Ag

Au

Pt

Ir

Rh

Os

Re

Mo

W

Cr

Mn

Fe

Co

Ni

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117 AND 118 COLUMNS		119 AND 120 COLUMNS	
PROCESSING AND PROPERTIES INDEX			
24-100. Aluminum Industry in Hungary. A. Domony and E. Kovacs. <i>Light Metals</i>. v. 10, July 1947, p. 333-344. A broad review of the growth and present status of the Hungarian alu- minum industry, emphasizing domestic applications.			
A 50-31A METALLURGICAL LITERATURE CLASSIFICATION			
117 AND 118 COLUMNS		119 AND 120 COLUMNS	
117 AND 118 COLUMNS		119 AND 120 COLUMNS	

.A. .

Distribution and Capacity of the World's Aluminum Producing Plants.
Léonard Doron. (Aluminium (Budapest), 1949, 1, (6), 17-22).--Shows by a
number of tables and maps the distribution, capacity, and the expansion
of aluminum production in various countries of the world.--S.3.

12 97 The bearing of metal impurities on furnace atmosphere and the influence of the technology of wet-drawing on the conductivity of light metals. By A. BERNARDI (<i>Aluminium</i> - Vol. I, No. 6, pp. 111-112, and No. 7, pp. 150-164, June-July, 1953). The author comments on the various aluminum alloys (Fe-Mn, Fe-Al, Fe-Mg, Fe-Al-V, Mn-Cr) and also on the conductivity of such alloys by wet-drawing. There is a need for a certain protection of impurities, but exceeding this point it will reduce sensitivity to a great extent in primary aluminum production mainly of aluminum and scandium. In the purity of aluminum, attention should also be given to the possible impurities that may be found in anodic current and in cathode carbons. Other metal-impurities present in domestic metals do not exceed 0.001 per cent. The second part of the article deals with the strength and conductivity of various aluminum alloys. This question has been successfully solved in producing an aluminum alloy capable of satisfying the requirements in strength, corrosion resistance and conductivity. This alloy belongs to the Al-Mg-Si type; the essence of which consists in the MgSi crystals formed by an appropriate heat treatment will arrange themselves into crystal structures, increasing their conductivity by not more than 10 per cent, while at the same time their strength increases by about 20 per cent. In the repetitive action of cold-chamber annealing, the degree of strengthening produced remains constant. Finally, the Al-Mg-Si alloy is described as being particularly satisfactory also to the requirements of casting. In conclusion, a very interesting article dealing with the Al-Mg-Si alloy is presented.	
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100-1000 MATERIALS F-000000	100-1000 METALS AND ALLOYS F-000000

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS																													
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B 17 Aluminium for Electrical Conductors. András Dancs. <i>Light Metals</i>, v. 12, Nov. 1949, p. 614-626. Translated from <i>Aluminium</i> (Hungary), no. 1, 1949, p. 139. Reviews relationship between the metallurgical and electrical properties of Al and certain Al alloys. Outlines production procedure for metal at the smelter and for conductor wire, and indicates current lines of research towards improving materials at present available for such wire. Data are tabulated.																																							
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ALUMINUM
VOL. II, 1950
NO. 1, JAN.

A. Dominy
Chemical polishing 18 20

ASIA-SLA METALLURGICAL LITERATURE CLASSIFICATION

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21

Uses of Aluminium in Food Preservation and Their Present and Future Problems. Andre Dony (Aluminium (Budapest), 1930, 2, (4), 89 (10)).— [In Hungarian]. Aluminium being readily available and in plentiful supply in Hungary, whereas tin is not always obtainable, great efforts are being made to substitute the former for tinplate and tinfoil in the canning and food-packing industries. In comparing the two materials, the merits and demerits of 99.8% aluminium and alloys containing 1.5% manganese and 1.5% magnesium are listed and briefly discussed. Particular attention is paid to problems connected with the pickling, anodizing, lacquering, and the airtight closing. D. is in favour of using chemical methods to obtain a suitable oxide coating and mentions the efforts being made in Hungary to apply the method of supersonic soldering to the last-mentioned problem. Factors influencing the corrosion of aluminium containers depend either on the container itself, e.g. cracks in the lacquer, faults in the oxide layer, impurities in the material, etc., or on the nature of the foodstuffs, e.g. their moisture content, pH (acidity), and the presence or absence of inhibitors such as sugar and fats. 101 references.—E. R.

more 1200

CA

Corrosion resistance of aluminum in ethyl alcohol
 Ambrós-Annay and Ilona Wokhauer (Magyar Alumínium
 Intézet, Budapest, Hung.) *Alumínium* 2
 175 (1950). -- Samples of an anticorrosive Al (A) contg. Cu
 0.08, Mn 0.74, Fe 0.20, Mg 0.54, and Si 1.08, of a soft one of
 refined Al (B) contg. Cu 0.04, Fe 0.18, and Si 0.16, and of a
 hard refined 99.5% Al (C) contg. Cu 0.05, Fe 0.30, and Si
 0.31% were kept 98 days in 98% EtOH at 20°. The av. wt. loss of
 liquid was stirred 45 days at 135° p.m. The av. wt. loss of
 the sheets was (A) 0.0063, (B) 0.0045, and (C) 0.0071
 g./sq. m./day. The tearing strength remained unchanged.
 When the EtOH concn. was 75, 40, 5, and 2%, the av. wt.
 loss of (B) was 0.019, 0.011, 0.021, and 0.043 g./sq. m./day.
 resp. With 100, 75, 40, 20, and 5% EtOH the av. wt.
 loss of (B) was 0.008, 0.015, 0.003, 0.011, and 0.003 g./sq. m.
 day, resp. For Al sheets as industrial construction material
 also high-purity grades (above 99.5%) should be used, and the
 natural oxide film should be reinforced by treatment with
 water glass, steam, varnish, plastic spray, etc. Micro-
 photos of Al plates with various protective coatings are
 given. István Fényes

A microfiche card with a header section containing the text "ALUMINIUM VOL. 3--1950/ No. 3, March". The main section of the card contains the title "Artificial strengthening of aluminium oxide film by chemical processes: a new way of forming protective coatings on light metals." followed by the number "62 63". The card is surrounded by a microfiche grid. On the right side, there is a small table of contents with the following entries: "CONTENTS" (1-2), "ALUMINIUM" (3-4), "ALUMINIUM OXIDE FILM" (5-6), "ALUMINIUM OXIDE FILM" (7-8), "ALUMINIUM OXIDE FILM" (9-10), "ALUMINIUM OXIDE FILM" (11-12), "ALUMINIUM OXIDE FILM" (13-14), "ALUMINIUM OXIDE FILM" (15-16), "ALUMINIUM OXIDE FILM" (17-18), "ALUMINIUM OXIDE FILM" (19-20), "ALUMINIUM OXIDE FILM" (21-22), "ALUMINIUM OXIDE FILM" (23-24), "ALUMINIUM OXIDE FILM" (25-26), "ALUMINIUM OXIDE FILM" (27-28), "ALUMINIUM OXIDE FILM" (29-30), "ALUMINIUM OXIDE FILM" (31-32), "ALUMINIUM OXIDE FILM" (33-34), "ALUMINIUM OXIDE FILM" (35-36), "ALUMINIUM OXIDE FILM" (37-38), "ALUMINIUM OXIDE FILM" (39-40), "ALUMINIUM OXIDE FILM" (41-42), "ALUMINIUM OXIDE FILM" (43-44), "ALUMINIUM OXIDE FILM" (45-46), "ALUMINIUM OXIDE FILM" (47-48), "ALUMINIUM OXIDE FILM" (49-50), "ALUMINIUM OXIDE FILM" (51-52), "ALUMINIUM OXIDE FILM" (53-54), "ALUMINIUM OXIDE FILM" (55-56), "ALUMINIUM OXIDE FILM" (57-58), "ALUMINIUM OXIDE FILM" (59-60), "ALUMINIUM OXIDE FILM" (61-62), "ALUMINIUM OXIDE FILM" (63-64), "ALUMINIUM OXIDE FILM" (65-66), "ALUMINIUM OXIDE FILM" (67-68), "ALUMINIUM OXIDE FILM" (69-70), "ALUMINIUM OXIDE FILM" (71-72), "ALUMINIUM OXIDE FILM" (73-74), "ALUMINIUM OXIDE FILM" (75-76), "ALUMINIUM OXIDE FILM" (77-78), "ALUMINIUM OXIDE FILM" (79-80), "ALUMINIUM OXIDE FILM" (81-82), "ALUMINIUM OXIDE FILM" (83-84), "ALUMINIUM OXIDE FILM" (85-86), "ALUMINIUM OXIDE FILM" (87-88), "ALUMINIUM OXIDE FILM" (89-90), "ALUMINIUM OXIDE FILM" (91-92), "ALUMINIUM OXIDE FILM" (93-94), "ALUMINIUM OXIDE FILM" (95-96), "ALUMINIUM OXIDE FILM" (97-98), "ALUMINIUM OXIDE FILM" (99-100).

ALUMINIUM
VOL. III.--I950.
No. 4, April.

A. Dominy.
Artificial strengthening of aluminium
oxide film by chemical processes; a
new way of forming protective coat-
ing on light metals. II

7A 11

ASB 514 METALLURGICAL LITERATURE CLASSIFICATION

CA

7

A few extraordinary cases of brass corrosion in an atmosphere containing sulfurous gases and investigation of such cases. Andráš Hunyady. *Bányász. Kohász. Lapok* 5 (1911), 421-5 (1910). When brass samples contg. 70, 63, or 58% Cu were kept 2 hrs. in an atm. satd. with SO_2 and steam and then stored for a week, the highest wt. losses were observed in brasses with the highest Cu contents. When similar brass samples were treated in an atm. contg. SO_2 , 1 or 11.5% at 450° , or in an atm. contg. SO_2 1% at 600° , the film formed on the surface of the brass contained Cu 70 and Zn 30, Cu 68 and Zn 32, Cu 10 and Zn 90, resp. for brass contg. Cu 70 and Zn 30; the film contained Cu 86 and Zn 14, Cu 36 and Zn 64, or Cu 2 and Zn 98, resp., for brass contg. Cu 63 and Zn 37; the film contained Cu 15 and Zn 85, Cu 30 and Zn 70, or Cu 0.0 and Zn 100, resp., for brass contg. Cu 58 and Zn 42%. When similar brass samples were kept at 20° in artificial sea water 2 weeks, the most severe corrosion occurred on the surface of brass contg. 58% Cu. The expts. show that the Cu content of brass alone is not the most important factor in corrosion resistance. Corrosion seems to depend on the quality of the protective coating formed under corrosive conditions. α -Brass is generally more resistant owing to its homogeneous structure. Since, however, strongly adhesive ZnO films are not readily formed on the surface of α -brasses, they are sometimes corroded more easily than brasses of the α -plus- δ -structure. Two opposing processes take place in brass corrosion.

István Finálv

Chemical Abst.
Vol. 48
Apr. 10, 1954
Metallurgy and Metallography

Demotz, Andras; Aluizinkum. Budapest: Népszava.
1953 (1954) 20 p.

CA

Polishing and finishing by chemical methods. Andras Keleny (Magyar Alumíniumkutató Intézet, Budapest, Hung.). *Aluminium* 3, 18 28 (1951). When articles of light metals or of Al are immersed in a liquid, cathodic processes take place on the prominent miniature peaks of the metal surface, whereas in the miniature valleys of the uneven surface anodic processes occur. Thus local differences arise in the concentration, and various pH values occur in different points in the liquid. When a liquid is adjusted to the proper pH value, then the prominent particles will be dissolved by this liquid, and at the same time a strongly adherent protective coating will be formed in the fine cavities of the surface. After a definite time the whole surface will be evenly polished. For this purpose liquids containing H_2PO_4 were most suitable. With baths at 85-90° a treatment of 3-10 min. is satisfactory. The articles are then washed in cold and hot water, dried carefully, and finished in a 2nd bath at 85-90° for 2-3 min. The best results were obtained when a mixture of H_2PO_4 and H_2SO_4 with some heavy metal salts as catalyst was used as the 1st bath and a water-free mixture of H_2PO_4 and H_2SO_4 with the same catalyst was used as the 2nd bath. The keeping quality of polished and finished surfaces prepared by this chemical method was extremely good owing to a fine, compact film of Al phosphate which prevents further oxidation. Chemically polished Al surfaces can be "electrodeposited" well. István Fényes

Met. Abstract

7

The Artificial Thickening of the Oxide Film on Aluminium by Chemical Means: New Method of Producing Protective Coatings on Light Metals. Andras Domony (Aluminium (Budapest), 1951, 3, (4), 73-77).--(In Hungarian). Theoretical and experimental data are given. The results of corrosion tests showed that specimens treated with the M.B.V. process (which is briefly described) suffered the least weight loss in a number of highly corrosive substances. 9 ref.--I. S. M.

Handwritten: J. J. A.
17A

A New Process for the Refining of Aluminium Scrap.
András Domany and R. Várhelyi (*Aluminium (Budapest)*,
1961, 8, (8), 183-192).—[In Hungarian]. After a brief
review of types of scrap and methods of refining them, and
an appraisal of the method of treating the melt with Cl_2 ,
the new H_2Cl_2 process is described. Thermodynamic aspects
and the practice of the process are shown, and special ref.
is made to the effect of H_2Cl_2 on the Mg content.—J. S. M.

Domony, A.

34. New possibilities for utilizing castings made from aluminium waste — Hullad-
ekaluminumból gyártottöntvények újabb felhasználási lehetőségei — by A. Domony
(Hungarian Engineering — Magyar Technika — No. 4, pp. 26-29, April 1951, 3 figs.,
1 tab.)

Contrary to other metals, aluminium waste occurring in the manufacturing processes cannot be utilized simply by remelting, therefore, up to now only third rate castings could be produced from this waste (fittings, door latches, stove legs, etc.); the actual ratio of the finished product to waste is 5:1. It is possible, however, to purify the waste metal according to the processes described in detail in the article to a degree that even high grade goods, such as engine parts, forgings subjected to high mechanical stresses, and, possibly, pistons and crank cases, etc. can also be produced. The various stages of purification are: the expulsion of the H_2 gas dissolved in the aluminium by means of chlorine gas, and, subsequently, by means of N_2 , then establishing a close contact between the contaminating metal bath and a layer of salt (e.g. of cryolite + $NaCl$ + KCl) with the addition of halogenides of heavy metals (e.g. $MnCl_2$ and $PbCl_2$) by which the mechanical properties of the material and the formation of its crystalline structure are favourably influenced.

DOMONY, A.

85. Artificial thickening of aluminum oxide film by chemical processes. CH

A new method of producing protective coating for light metals - Az aluminium felületi oxidációjának kémiai módszerekkel történő megerősítése; keményfémek festékalapjának újfajta kiképzése - by A. Domony (Aluminium - Vol. III, No. 3, 4, pp. 62-65, 73-77, March, April 1951, 15 figs., 3 tabs.)

Experiments were made with the object of finding a process by which the favourable properties of artificially thickened oxide films will ensure an efficient corrosion protection and will satisfy the requirements for a suitable coating that can be used to advantage in the treatment of large workpieces without having to apply dip baths. Thorough investigations on the formation as well as the chemical and physical properties of aluminum oxide films led to the conclusion that new type artificial oxide films can be prepared efficiently by spraying an aqueous solution of potassium dichromate, sodium carbonate and solid sodium hydroxide. This solution can be sprayed quickly and easily on aluminum surfaces, and forms a corrosion resistant spongy oxide film with good varnish absorbing and adhesive properties. Oxidation sets in at about 30 C° and becomes very intensive at 50 to 60 C°. Various types of varnishes can be readily applied to aluminum surfaces prepared in this manner resulting in satisfactory adhesion.

Effect of boron on the electrical conductivity of reduced and ordinary aluminum. A. Domany and E. K. Vaynshteyn (Research in Nonferrous Metals, Moscow, U.S.S.R. Acad. Sci. Div. Phys. Sci., 1959-60, 1960) (in Russian).-- The sharp lowering of the elec. cond. of Al by Ti (and V) can be cancelled to a large extent by alloying with boron. The elec. cond. of pure Al (Fe + Si < 0.1%) with boron added up to 1% does not fall more than 0.5 m./sq. cm. The curve for Ti plus B (where B/Ti > 1) was 1.5 m./sq. cm. lower and parallel to the characteristic curve of the metal alloyed only with boron. The elec. cond. of ordinary Al contg. Ti is similarly improved, but the effects of Fe and Si are unchanged. Titanium boride is presumably formed. R. D. Mith

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Domony, (T)

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B. T. K.
V. 3 No. 3
Mar. 1954
Metals- Extraction
And Refining

3662* Production and Properties of Alloying Materials
Necessary for the Manufacture of ~~Aluminum~~ Alloys. (Hungarian.) ~~Andras Domony and Károly Várhegyi~~
Aluminium (Budapest), v. 3, no. 11, Nov. 1953, p. 238-241.
Places emphasis on the importance of keeping out Fe. Practical
details of various processes are outlined. Micrograph, tables,
graphs. 10 ref.

B. T. R.
Vol. 3 No. 4
Apr. 1954
Pollution and Wastes

5656* Utilization of Salt Residues and Scrapings Arising During the Processing of Light Metals and Light Metal Scrap. (Russian.) Andreev, I. *Metallurgiya* (Moscow), 1954, no. 12, Dec. 1953, 12-13, 14-15, 16-17, 18-19, 20-21, 22-23, 24-25, 26-27, 28-29, 30-31, 32-33, 34-35, 36-37, 38-39, 40-41, 42-43, 44-45, 46-47, 48-49, 50-51, 52-53, 54-55, 56-57, 58-59, 60-61, 62-63, 64-65, 66-67, 68-69, 70-71, 72-73, 74-75, 76-77, 78-79, 80-81, 82-83, 84-85, 86-87, 88-89, 90-91, 92-93, 94-95, 96-97, 98-99, 100-101, 102-103, 104-105, 106-107, 108-109, 110-111, 112-113, 114-115, 116-117, 118-119, 120-121, 122-123, 124-125, 126-127, 128-129, 130-131, 132-133, 134-135, 136-137, 138-139, 140-141, 142-143, 144-145, 146-147, 148-149, 150-151, 152-153, 154-155, 156-157, 158-159, 160-161, 162-163, 164-165, 166-167, 168-169, 170-171, 172-173, 174-175, 176-177, 178-179, 180-181, 182-183, 184-185, 186-187, 188-189, 190-191, 192-193, 194-195, 196-197, 198-199, 200-201, 202-203, 204-205, 206-207, 208-209, 210-211, 212-213, 214-215, 216-217, 218-219, 220-221, 222-223, 224-225, 226-227, 228-229, 230-231, 232-233, 234-235, 236-237, 238-239, 240-241, 242-243, 244-245, 246-247, 248-249, 250-251, 252-253, 254-255, 256-257, 258-259, 260-261, 262-263, 264-265, 266-267, 268-269, 270-271, 272-273, 274-275, 276-277, 278-279, 280-281, 282-283, 284-285, 286-287, 288-289, 290-291, 292-293, 294-295, 296-297, 298-299, 300-301, 302-303, 304-305, 306-307, 308-309, 310-311, 312-313, 314-315, 316-317, 318-319, 320-321, 322-323, 324-325, 326-327, 328-329, 330-331, 332-333, 334-335, 336-337, 338-339, 340-341, 342-343, 344-345, 346-347, 348-349, 350-351, 352-353, 354-355, 356-357, 358-359, 360-361, 362-363, 364-365, 366-367, 368-369, 370-371, 372-373, 374-375, 376-377, 378-379, 380-381, 382-383, 384-385, 386-387, 388-389, 390-391, 392-393, 394-395, 396-397, 398-399, 400-401, 402-403, 404-405, 406-407, 408-409, 410-411, 412-413, 414-415, 416-417, 418-419, 420-421, 422-423, 424-425, 426-427, 428-429, 430-431, 432-433, 434-435, 436-437, 438-439, 440-441, 442-443, 444-445, 446-447, 448-449, 450-451, 452-453, 454-455, 456-457, 458-459, 460-461, 462-463, 464-465, 466-467, 468-469, 470-471, 472-473, 474-475, 476-477, 478-479, 480-481, 482-483, 484-485, 486-487, 488-489, 490-491, 492-493, 494-495, 496-497, 498-499, 500-501, 502-503, 504-505, 506-507, 508-509, 510-511, 512-513, 514-515, 516-517, 518-519, 520-521, 522-523, 524-525, 526-527, 528-529, 530-531, 532-533, 534-535, 536-537, 538-539, 540-541, 542-543, 544-545, 546-547, 548-549, 550-551, 552-553, 554-555, 556-557, 558-559, 560-561, 562-563, 564-565, 566-567, 568-569, 570-571, 572-573, 574-575, 576-577, 578-579, 580-581, 582-583, 584-585, 586-587, 588-589, 590-591, 592-593, 594-595, 596-597, 598-599, 600-601, 602-603, 604-605, 606-607, 608-609, 610-611, 612-613, 614-615, 616-617, 618-619, 620-621, 622-623, 624-625, 626-627, 628-629, 630-631, 632-633, 634-635, 636-637, 638-639, 640-641, 642-643, 644-645, 646-647, 648-649, 650-651, 652-653, 654-655, 656-657, 658-659, 660-661, 662-663, 664-665, 666-667, 668-669, 670-671, 672-673, 674-675, 676-677, 678-679, 680-681, 682-683, 684-685, 686-687, 688-689, 690-691, 692-693, 694-695, 696-697, 698-699, 700-701, 702-703, 704-705, 706-707, 708-709, 710-711, 712-713, 714-715, 716-717, 718-719, 720-721, 722-723, 724-725, 726-727, 728-729, 730-731, 732-733, 734-735, 736-737, 738-739, 740-741, 742-743, 744-745, 746-747, 748-749, 750-751, 752-753, 754-755, 756-757, 758-759, 760-761, 762-763, 764-765, 766-767, 768-769, 770-771, 772-773, 774-775, 776-777, 778-779, 780-781, 782-783, 784-785, 786-787, 788-789, 790-791, 792-793, 794-795, 796-797, 798-799, 800-801, 802-803, 804-805, 806-807, 808-809, 810-811, 812-813, 814-815, 816-817, 818-819, 820-821, 822-823, 824-825, 826-827, 828-829, 830-831, 832-833, 834-835, 836-837, 838-839, 840-841, 842-843, 844-845, 846-847, 848-849, 850-851, 852-853, 854-855, 856-857, 858-859, 860-861, 862-863, 864-865, 866-867, 868-869, 870-871, 872-873, 874-875, 876-877, 878-879, 880-881, 882-883, 884-885, 886-887, 888-889, 890-891, 892-893, 894-895, 896-897, 898-899, 900-901, 902-903, 904-905, 906-907, 908-909, 910-911, 912-913, 914-915, 916-917, 918-919, 920-921, 922-923, 92

80. Latest view points and corrosion data on the utilization of aluminum alloys for shipbuilding. - A. D. G. M. A. H. 116. (Journ. de Geph. - Vol. 1, 1954, No. 11, pp. 349-355, No. 12, pp. 366-373, 18 figs., 6 tabs.)

Aluminum alloys and different grades of steel produced in Hungary for use in shipbuilding were subjected to comparative tests for their resistance to corrosion. The following tests were applied: (1) testing in liquid (stationary and flowing) medium, (2) stress corrosion test (static and dynamic loads), (3) salt spray chamber test. It was established that the tested aluminum alloys were less sensitive to the corrosion by the substances occurring in ship operation than were steel plates. There was no substantial difference between the corrosion resistance of the various copper-free aluminum alloys. It was found that the use of aluminum alloys with a copper content should be avoided in the building of vessels both for inland navigation and seafaring. Al-Mg type alloys showed the best resistance to corrosion.

However, the tendency towards stress corrosion increased rapidly with the increase of the Mg content. From among the Al-Mg-Si type alloys the corrosion resistance of annealed and solution treated materials was the best. Thus it was proved that aluminum alloys could be successfully used in shipbuilding, a fact verified in practice by the watercraft built in Hungary for inland navigation.

DOMONY, Andras, a muszaki tudomanyok kandidatusa; VARHELYI, Rezso

Effect of the casting skin of aluminum billets cast in water
on the properties of sheets made of the aluminum billets.
Koh lap 9 no. 11: 505-508 N '54.

1. Femipari Kutato Intezet.

Domony, A

73. Influence of some alloying constituents on the structure of cast aluminium. (In English) A. Domony, K. R. Vassal. *Acta Technica Academiae Scientiarum Hungaricae*, Vol. 12, 1955, No. 1—2, pp. 157—164; 1 tab.

Experiments were conducted to show the effect of the various alloying elements and impurities of aluminium on the primary grain size of the cast metal. Grain size and mode of crystallization depend to the greatest extent — according to the experiments conducted under the same melting and casting conditions — on the kind of alloying constituents present in the molten metal, on the presence of other metallic constituents or impurities and on the amount. As anticipated, titanium showed an effect on grain size which was higher by several orders than that of the other investigated elements. In the opinion of the author, the increased grain refining effect and its influence on crystallization cannot be explained univocally by any of the following theories: peritectic reaction, nuclei of titanium carbide particles and remnant crystal lattice fragments. A much clearer explanation can be given by assuming a favourable equilibrium of the forces and energies acting between the aluminium and titanium atoms in a melt of approx. 700° C.

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DOMONY, A.

Some current problems with our finished aluminum products. p. 21. KOHASZATI
LAPOK. (Magyar Banyaszati es Kohaszati Egyesulet) Budapest. Vol. 10, no. 1,
Jan. 1955.

SOURCE: East European Accessions List (EEAL), Library of Congress
Vol. 5, no. 6, June 1956

DOMONY, A.; VASSEL, R.; VARHELYI, R.

DOMONY, A.; VASSEL, R.; VARHELYI, R. Experiments for eliminating abnormalities in
electric conductivity of aluminum wires. p. 497.

Vol. 15, No. 1/4, 1955.

NYCZLEMENYEI.

TECHNOLOGY

Budapest, Hungary

So: East European Accession, Vol. 5, No. 5, May 1956

2111* (Hungarian) Abandonment of Internationally Accepted
Standard Quality. Hungarian Production; Results in 1955
Nemzetközi előírásoknak megfelelő alumíniumvezetékek
1955 évi hazai gyártásának tapasztalatai. Árvács Dömény,
Robert Vassel, and János Dery. *Kutatói Lapok* V. 11. no. 8,
Aug. 1956, p. 366-67.

Review of laboratory and industrial experiments for the im-
provement of the properties of electrically conductive Al wire.
Effects of production methods on properties of wire.

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Relation between the characteristics of the hardening curve of electrical aluminum wire and the composition of the metal. András Domonyi, Fehér Katalin, Keszthelyi 1956, 176-81. A comparison of the hardening curves of Al wires of different compos. with addns. of Fe (0.003-0.43, Si 0.002-0.13, Ti 0.002-0.015, V <0.005-0.020, and B 0.01% shows that the character of the curves is detd. by the distribution of the impurities rather than by the type of alloy. Felicitas D. Goodman

ANDRAS DOMONY

Large-scale production of aluminum wire of good conductivity. Andras Domony and K. Robert Vancs. *Fluoridat Kutato* *Intell. Kozlismayei* 1956, 183-84. The elec. properties of the finished wire depend upon the compn. of the basic metal and the manufg. conditions. The Si content of the Al must be $\leq 0.1\%$. During manuf. the formation of the Al-Si solid soln. must be avoided. The deleterious effect of the V and Ti content can be offset by addins. of B.

Felicitas D. Goodman

DOMONY, A.

Effect on the technological properties of aluminum of small quantities of alloy elements.
p.185. (Kohaszati Lapok. Budapest. Vol. 11, no. 4, Apr. 1956.)

SO: Monthly List of East European Accessions (EEAL) LC., Vol. 6, no. 7, July 1957 Uncl.

*Light Metal Congress Organized by the Hungarian Academy
of Sciences, Sept. 28-30, 1951, Budapest. A. Demeny (Acta
Techn. Acad. Sci. Hungar. 1952, 14, 451, 571-574).*

method of production of new iron-
reduction plants, (4) the ironing of new iron-
ferrous metallurgy, and (5) the economic use of Al. Lectures
were devoted to: (i) problems related to raw materials for
producing light alloys; (ii) the metallurgy of light alloys,
and (iii) the metallography and technology of light alloys.
— *Delezi* considered that the best conditions for the
electrolytic prodn. of Al are provided by the presence of 1%
in the electrolyte. — *Petrov* summarized his recent work on
the development of new light alloys for the construction of
supersonic aircraft, and dealt with the prodn. of very high-
purity Al, and segregation during chill-casting of Al. *A.
Delezi* reported on his theoretical calculations and practical

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Donny R.
 results relating to the form operations in the rolling, extruding, wire- and deep-drawing of Al and its alloys. J. Seky gave new explanations of the structural changes occurring during the hot-rolling of Al; he considered that no recrystallization occurs during the process. A. Domany thought that the effect of individual impurities on the technological properties of Al depend, *(inter alia)*, on the form in which these impurities are present and how they affect each other. -- *Paperny* reported on the metallographic and technological characteristics of pitons forged from an Al-Cu-Ni-Mg alloy containing small proportions of Si and Mn. F. Erdmann-Jennazy summarized his researches relating to the weldability of Al-Mg, Al-Mn, and Al-Mg-Si alloys. -- *(Lidars)* reported on the development of arc-welding under Ar protection, stressing the results obtained with a process of automatic welding, using D.C. and vertical electrodes, suitable for welding thick plates, with a welding speed of 1-2 in./sec. Transfer of metal from the electrode to the metal pool, in the case of pure Al occurs by a high-speed, almost instantaneous migration of metal drops; in the case of Al alloys, especially alloys of the Al-Mg type, the drops do not form a continuous stream, and this favours the formation of gas pockets. -- *Senicki* called special attention to "Pitoveru" alloy, an Fe alloy containing ~30% Al. Particulars of these papers and others submitted to the Congress are contained in the report. The Congress is to be held annually in different countries. — J. S. Q. T.

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DOMONY, A., VASSEL, R.

Effect of some alloys on the texture of cast aluminum. P. 17
KOZLEMEYEI Budapest, Vol. 18, no. 1/4, 1956

SOURCE: East European Accessions List (EEAL) Library of Congress
Vol. 5, no. 8, August 1956

DONOHY, A.

A survey of the situation of the bauxite, alumina, and aluminum industries of the world and its effect on the Hungarian aluminum industry. (To be contd)p. 130. (KOVASZATI LAPOK. Vol. 12, no. 3, Mar. 1957, Budapest, Hungary)

SO: Monthly List of East European Accessions (EEAL) LC. Vol. 6, no. 12, Dec. 1957.
Uncl.

DOMONY, Andras, dr., a muszaki tudomanyok kandidatusa

Survey of the world's bauxite, alumina and aluminum industries and their effect on the Hungarian aluminum industry. Koh lap 12 no. 4/5:176-180 Ap-May '57.

1. "Kohászati Lapok" szerkeszto bizottsagi tagja.

18 6 1
 Effect of small quantities of alloying elements and impurities on some physical properties of aluminum. Aluminum (Pechenkin, metallurg. Pribl.) (in Russian). Metall. Sci. Hung. 16, 103-81 (1957) (in German). — The elec. cond. of Al depends mainly on the quantity of the dissolved elements in the matrix, of such as Si, Ti, and V. Small quantities of dissolved Si can be removed from the solid soln. by suitable heat-treatment, while dissolved Ti or V can be removed by the addn. of B. The min. values of the plots tensile strength against reduction by cold-working depend on the quantity of the impurities present, and the character on the distribution of the alloying elements within the metal. The character of the strain-hardening curves of Al, contg. small amts. of segregated elements, is the same as that of the high-purity Al. The character of the strain-hardening curve of Al, contg. some 0.01-0.1% alloying elements in solid soln. is somewhat different. The alteration of the character of the strain-hardening curves might be explained with the help of the theory of dislocations.
 Frederick C. Nash[il]

See
 1855. Monthly List of East European
 Accessions (EEAAL) LC, Vol. 6, No. 8,
 Aug. 57 — incl —

DOMONY, Andras, dr., a muszaki tudomanyok kandidatusa

The state and development of the industry of finished aluminum products from the point of view of their application in Hungary. Koh lap 12 no. 11/12 481-488 N-D '57.

Some newer experimental results in aluminum corrosion under the operational conditions corresponding to those of nuclear reactors. Ibid.:542-550.

1. "Kohasziati Lapok" szerkeszto bizottsagi tagja.

HUNGARY / Chemical Technology. Chemical Products and H-4
Their Application. Corrosion. Corrosion
Control.

Abs Jour: Ref Zhur-Kimiya, No 1, 1959, 1631.

Author : ~~Domony, A.~~
Inst : Not given.
Title : Investigation of Corrosion Aluminum Under Con-
ditions of an Operating Atomic Reactor.

Orig Pub: Kohash. lapok, 1957, 12, No 11-12, 542-550.

Abstract: The corrosion resistance of aluminum with addit-
ions of Fe, Ti, Si and B in water was investigated
at a high pressure and temperature. It was estab-
lished that the alloys of Al, containing inter-
metal compounds have the best corrosion resistance.
By the effect of the latter (depolarizing proper-
ties) can be explained the very high corrosion re-
sistance of super-pure aluminum in a weakly aggres-
sive media.

Card 1/1

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HUNGARY/Chemical Technology. Chemical Products and Their Applica- H-4
tion. Corrosion, Corrosion Control

Abs Jour : Ref Zhur - Khim., No 24, 1958, No 81963

Author : Domony A., Lichtenberger E.

Inst :

Title : New Approach to the Corrosion Problems of Aluminum in the
Weakly Aggressive Media and in Food Products

Orig Pub : Kohasz. lapok, 1957, 12, No 11-12, 550-554

Abstract : Corrosion resistance of Al-strips, anodized in H_2SO_4 solu-
tion and in a mixture of H_2SO_4 and CH_3COOH , was investiga-
ted. It was concluded that the best corrosion resistive
properties have Al-strips that were chemically oxidized.
This is explained by the structure of the protective layer.
Thickness and structures of various oxide layers were
investigated.

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Distr: 4E2c/4E3c/4E3d

18 27
Corrosion of aluminum under conditions encountered in nuclear reactors. (Andras Domony. *Kohászati Lapok* 90, 542-55 (1957).—Page 54 and 55 contg. various units. of ²Fe, Si² and (or) Ti were subjected to 1-8 cycles consisting of 8 hrs. exposure to steam at 180° and cooling for 16 hrs. The exposed samples were examd. for tensile, microscopical, x-ray, and chem. properties. Al contg. small amts. of Si and Ti will corrode in relation to the temp. of the aq. phase; at 180° corrosion (I) begins after a few hrs. incubation, at >260° within a few min.; >0.2% Fe will substantially inhibit I at 180° (>0.3% at 260°); >0.4% Si, if no Fe is present, will prevent I; 0.2% Ti will be effective at 180° but not at 260°. B will accelerate I unless Si is also present, in the latter case it shows no appreciable effect. I proceeds inwards along the crystal boundaries and is attributed to the water (which is dissolved at high temps. and pressures) in that the OH ions will peptize the oxide layer at the surface which, in the absence of air, is unable to be regenerated. The depolarizing effects of the various ingredient metals, depending upon the potential differential, are of major effect in inhibiting or accelerating I. L. G. Arxat.

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DEMENY, A.

HUNGARY/Chemical Technology - Chemical Products and Their
Application. Food Industry.

H.

Abs Jour : Ref Zhur - Khimiya, No 10, 1959, 36886

Author : Domony, A.

Inst : --

Title : New Opportunities of the Utilization of Aluminum in the
Production of Equipment for the Food Industry and for
the Packing of Food Products.

Orig Pub : Elekt. Izv., 1958, 12, No 8-9, 235-245.

Abstract : The properties of aluminum and its alloys, especially
from the point of view of corrosion in the process of
food production, were examined. By prolonged boiling
in distilled water, a protective film is formed on the
aluminum surface, which in ordinary conditions protects
the equipment from corrosion. Other methods of protec-
tive treatments of aluminum surfaces were described.
The expansion in the production of packing equipment

Card 1/2

Distr: 4E2c

161. The effect of different impurities on the corrosion resistance of super-purity aluminium in water of high temperature and high pressure. (In German) A. Domon y: *Acta Technica Academiae Scientiarum Hungaricae*, Vol. 21, 1958, No. 1-2, pp. 123-130, 8 figs., 2 tabs.

The experiments showed the corrosion resistance of super-purity aluminium alloyed with various amounts of Si, Fe, Ti and H against water of high pressure and temperature. It was established that only aluminium containing certain kinds of foreign phase constituents in definite amounts is resistant to water of high pressure and temperature. Evaluating the results of the experiments on the basis of theoretical considerations, correlations were successfully established which explain the thus far unexplained unusual corrosion behaviour of super-purity aluminium. Accordingly, the corrosion behaviour of aluminium in mild agents is strongly influenced not only by the hitherto known factors (the formation of oxide films, the electrochemical behaviour of the different phases against the basic metal) but by the depolarizing effect of the foreign phases as well.

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Distr: 4E2c

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Distribution of calcium impurities in aluminum and their effect on the electrical conductivity of the metal. A. Dumony (Research Inst. Ind. Non-Ferrous Metals, Budapest). *Acta Tech. Acad. Sci. Hung.* 21, 295-307 (1968) (in German). —Melts of Al metal with CaF_2 , CaCl_2 , or CaI_2 contg. cryolite or chiolite salts can absorb finely dispersed impurities of nonmetallic salts that tend to promote gas absorption. No Al-Ca alloys are formed under these conditions (950° to 1150°). Ca found in Al metal indicates the presence of salt inclusions. These are responsible for decreased elec. conductivities, as well as for difficulties encountered in the degassing of the metal, but they can be dissolved out of the metal with mixts. of chiolite (rather than higher-melting cryolite) and NaCl at 720° to 800° . Alloying of Al metal with small amts. of Ca does not affect the gas absorption.

Hans U. D. Wiesendanger

Distr: 4E20

V Surface treatment of aluminum. André Demony, René Lichtenberger-Bajza, and Pat Cochin. *Korrosion* 13, 92-100(1958); *Metallberichter* 13, 84-7(1959).— Sheets made from pure Al (contg. 0.00, 0.10, and 0.28% Ca, resp.), cran.-grade Al, Al-Zn-Mg, Al-Zn-Mg, anticorrodal (Al contg. Fe 0.22, Si 0.74, Mg 0.52, and Mn 0.40%), masil (Fe 0.23, Si 1.21, Mg 1.05, and Mn 0.55%), nantal (Fe 0.22, Si 4.20, and Mg 0.25%), and dural (Fe 0.25, Si 4.05, Mg 0.70, and Mn 0.30%) were boiled 3 hrs. in the following baths: distd. water, tap water, distd. water with 2% H₂O₂, tap water with 2% H₂O₂, distd. water with 1% NH₄OH, tap water with 1% NH₄OH, distd. water with 2% NaNO₂, tap water with 2% NaNO₂, distd. water with 2% glycerol, tap water with 2% glycerol. The protective layers developed by these treatments were examd. for thickness, elec.-capacity properties, and resistance to corrosion. Results were given in detail. Improvements in the protective properties of these layers were effected by anodic oxidation by using a spray gun. It was found that the effectiveness of the treatment is influenced by the method used in the manuf. of the sheets.

L. G. Arval

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LOMONY, A.
LICHTENBERGER, E.
CSOKAN, P.

Some recent practical results in surface finishing of aluminum. p. 92.

KOHASZATI IAFOK. (Magyar Banyaszati es Kohaszati Egyesulet) Budapest, Hungary
Vol. 14, no. 2/3, Feb./Mar. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 8, August 1959
uncla.

H/014/60/000/002/001/003
E190/E435

18.12.10

AUTHORS: Buray, Zoltán, Doctor, Candidate of Technical Sciences
and Domony, András, Doctor, Candidate of Technical
Sciences

TITLE: The Corrosion of Aluminium Alloys and of Their Joints
as Used in Shipbuilding

PERIODICAL: Kohászati lapok, 1960, No.2, pp.56-60

TEXT: The corrosion-resistance of aluminium alloys, and
especially that of Al-Mg alloys, is well known in unstressed
conditions but few data are available on the importance of
Mg-content and of methods of heat-treatment on the corrosion-
resistance of 3 to 5% Mg alloys to more aggressive media under
simultaneous application of external stresses. The present study
aims at establishing the optimum composition and condition of
alloys in laboratory and field tests, with special reference to
corrosion in Danube and sea-water. Al-4%Cu-1.5%Mg,
Al-1.5%Mg-0.8%Si, Al-1.2%Mg-0.7%Si-0.5%Zn ("MASZIL"), Al-3%Mg,
Al-5%Mg, Al-2%Mg-1%Mn and Al-1%Mn sheets (in various conditions
of heat treatment), Al-1.7%Cu-1%Ni-0.5%Mg wires and cast

Card 1/5

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E190/E435

The Corrosion of Aluminium ...

Al-1%Mg-2.2%Si, Al-3%Mg and Al-9%Si test pieces were subjected to laboratory corrosion tests and the loss of weight, of tensile strength and of elongation were measured. Stirring tests in Danube-water gave the following order of merit (loss of strength and elongation increasing in the series):

1. Al-3%Mg, Al-5%Mg, Al-Mg-Mn, annealed Al-Mg-Si
2. Solution treated Al-Mg-Si
3. Solution treated and precipitated Al-Mg-Si
4. Al-Cu-Mg.

Synthetic sea-water gave the following order:

1. Al-3%Mg, Al-5%Mg
2. Solution treated Al-Mg-Si
3. Annealed Al-Mg-Si and MASZIL.

In salt-spray tests, alloys containing copper suffered 0.20 to 0.30 g/m² loss of weight in 30 days against only 0 to 0.03 g/m² of the other alloys (for comparison, steel lost 3.0 g/m² in the same period). Static tests in battery-acid (38.5% sulphuric acid) again showed the inferiority of alloys containing Cu whilst the other alloys gave more or less equal resistance. It is concluded that, in the absence of stresses, there is no significant difference in

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the chemical resistance of alloys containing 3% and 5% Mg. Tests in Danube-water with simultaneous application of external stresses, proved Al-Mg alloys to be far superior to any other alloys, including the heterogeneous Al-Mg-Si alloy that showed a good resistance in the unstressed condition. Tests in the more aggressive 3%NaCl + 1%HCl medium very clearly demonstrated the higher resistance of the Al-3%Mg alloy and the susceptibility of Al-5%Mg alloy in the heterogeneous condition. "Nautal" samples (4.25%Mg, 0.21%Fe, 0.27%Si, 0.22%Mn) were corrosion resistant even in the artificially aged condition (heated at 350°C for 3 hours followed by slow cooling and ageing at 150°C for 8 hours) thus, this alloy represents the maximum harmless Mg concentration. Further investigation concentrated on laboratory stirring-corrosion tests of "Nautal" and its butt-welded joints prepared by gas or argon-arc welding with tungsten (TIG) or metal (MIG) electrode. The progress of corrosion was followed by measuring loss of weight, thickness, strength, proof stress and elongation. No signs of corrosion were detected after 60 days of exposure to Danube-water on any of the specimens and the effect of synthetic sea-water (made with

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The Corrosion of Aluminium ...

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27.0 g NaCl, 3.8 g MgCl₂, 1.7 g MgSO₄, 1.2 g CaSO₄, 0.9 g K₂SO₄, 0.1 g CaCO₃ and 0.1 g MgBr₂ for each litre of water) was restricted to local greyish spots. In a 3%NaCl + 1%HCl solution, corrosion pits formed first adjacent to the welded bead and only later in the rest of the specimen. Weight loss was slow in the first 30 days but accelerated afterwards. The transition zone, adjacent to the bead, suffered greatest loss in weight and thickness. This zone was widest in the gas- and narrowest in the TIG-welded specimens. Because of this local decrease in thickness, tensile tests gave a specially sensitive indication of corrosion. Battery acid (with 38.5% H₂SO₄) attacked the specimens from the moment of immersion. In agreement with observations made in the hydrochloric acid test, the zone affected by moderate heat (200 to 500°C) during welding exhibited approximately twice the corrosion-resistance of the parent metal, whilst rapid corrosion in the transition-zone proper led to the formation of deep grooves. In order to confirm the laboratory tests, Al-3%Mg, "Nautal" and Al-Mg-Si specimens were fixed (with aluminium and hot-dip galvanized steel rivets) to a PVC sheet at the inner side of the bulwark at the bow of a Danube-sea going ship and were inspected for corrosion after a

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The Corrosion of Aluminium ...

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half-year cruise on the Black Sea and the Mediterranean Sea. No significant loss of strength was observed on any of the specimens. "Nautal" showed minimum surface attack; corrosion occurred only where rust had flown down from damaged steel rivets. All aluminium-riveted joints were intact except for a slight bloom where a gap was left in riveting. From the results of laboratory and field tests, it is concluded that "Nautal" represents the optimum composition for the building of welded and riveted load-bearing structures exposed to slightly corrosive atmospheres. The alloy is slightly easier to fabricate and to weld than Al-5%Mg, it resists corrosion by Danube- and sea-water, and joints welded under a protective atmosphere approach the parent metal in resistance to corrosion by hydrochloric and sulphuric acid. There are 16 figures, 4 tables and 3 Soviet-bloc references. ✓

[Abstractor's note: The first part of this article is published in No.1, pp.6-11; this is an abstract of the complete article.]

Card 5/5

DOMONY, Andras, dr.

Technical and economical description of more important methods for the surface treatment of light metals. Gepyartastechn 1 no.7:275-280 0 '61.

1. Research Institute of the Metal Industry, Budapest.

JAGANNADHA RAJU, G.J.V. (Budapest XI., Fehervari ut 144); DOMONY, A.
(Budapest XI., Fehervari ut 144)

The chemical resistance of aluminum materials of different composition
in concrete, mortar and gypsum. Periodica polytechn eng 5 no.2:117-133
'61.

1. Fehervari Kutato Intezet.

37963

S/137/62/000/005/027/150
A006/A101

5.2100

AUTHORS: Domony, A., Laár, T.

TITLE: Joint effect of titanium and boron upon the properties of an oxide film formed on an aluminum surface, and upon aluminum crystallization

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 5, 1962, 14, abstract 5082 ("Kohász. lapok", 1961, v. 94, no. 11, 505-507, Hungarian; Russian, German, English summaries)

TEXT: Some admixtures contained in Al, affect considerably its properties. Ti impairs electric conductivity, but promotes grain refining. B has a low effect on electric conductivity but promotes also the production of a fine crystalline structure. B impairs the properties of a surface oxide film. In the case of the joint presence of Ti and B, Al properties are determined by the predominance of one of the aforementioned elements in free state. To improve Al structure, 0.01 - 0.032% Ti should be added.

[Abstracter's note: Complete translation]

B. Gulyanitskiy

Card 1/1

DOMONY, Andras, a muszaki tudományok doktora

Reduction of corrosion damages as reflected in Hungary's
technical progress. Magy tud 69 no.3:167-171 Mr '62.

1. Tudományos főmunkatárs, Fémipari Kutató Intézet.

DOMONY, Andras, dr.

Society news. Koh lap 95 no.6:260 Je '62.

1. "Kohaszati Lapok" szerkeszto bizottsagi tagja.

HORVATH, Gyorgy; LATINAK, Istvan; DOMONY, Andras, dr.; OVARI, Antal

Society news. Koh lap 95 no.8:342 Ag '62.

1. "Kohasziati Lapok" szerkeszto bizottsagi tagja (for Domony and Ovari).

BURAY, Zoltan, dr., a muszaki tudományok kandidátusa; DOMONY, Andras, dr.,
a muszaki tudományok kandidátusa

Corrosional behavior of aluminum alloys and their joints used in
shipbuilding. (To be contd.). Koh lap 93 no.1:6-11 Ja '60.

1. "Kohászati Lapok" szerkesztő bizottsági tagja (for Domony).

BURAY, Zoltan, dr., a muszaki tudományok kandidátusa; DOMONY, Andras, dr.,
a muszaki tudományok kandidátusa

Corrosional behavior of aluminum alloys and their joints used in
the shipbuilding. (Continuation). Koh lap 93 no.2:56-60 F '60.

1. "Kohászati Lapok" szerkeszto bizottsagi tagja (for Domony).

SZELESS, Laszlo; DOMONY, Andras, dr.

The editorial policy of "Kohaszati Lapok." Koh lap 93 no.7:329-331
Jl '60.

1. "Kohaszati Lapok" szerkeszto bizottsagi tagja.

DOMONY, Andras, dr., a muszaki tudományok kandidátusa

Explaining the causes for some irregularities observed during aluminum processing. Koh lap '93 no.9:400-407 S '60.

1. "Kohászati Lapok" szerkeszto bizottsagi tagja.

ARKOS, Frigyes; DOMONY, Andras, dr.

Periodical reviews. Koh lap 93 no.11:526-527 N '60.

1. "Kohaszati Lapok" főszerkesztője (for Arkos);
2. "Kohaszati Lapok" szerkesztő bizottsági tagja (for Domony).

DOMONY, Andras, dr., a muszaki tudományok kandidátusa

Preparation of light-metal semifinished products especially suited qualitatively for chemical surface refinement. Koh lap 93 no.12: 570-575 D '60.

1. "Kohászati Lapok" szerkesztő bizottsági tagja.

DOMONY, Andras, dr.

Some specific cases of the behavior of light metals in
connection with corrosion. Gepgyartastechn 2 no.5:181-184
My '62.

1. Femipari Kutato Intezet.

DOMONY, Andras, dr.

Aluminum conference. Koh lap 96 no.4:Suppl: Ontode 14 no.4:3 of cover
Ap '63.

1. "Kohassati Lapok" szerkeszto bizottsagi tagja.

LAAR, Tibor, tudományos munkatárs; DOMONY, András, a muszáki tudományok
doktora

Developing the casting method in the foundries of aluminum
smelteries. Koh lap 96 no.5:232-236 My '63.

1. Fémipari Kutató Intézet (for Laar).
2. "Kohászati Lapok" szerkesztő bizottsági tagja (for Domony).

PITTER, Pal; ARKOS, Frigyes; HORVATH, Antal; DOMONY, Andras, dr.;
LEVARDI, Ferenc, dr.; SELMECI, Bela; FEKETE, Sandor; MARTOS,
Ferenc, dr.; MACSAY, Jozsef, okleveles gepeszmernok;
TARCHY-HORNOCH, Antal, dr., akademikus, egyetemi tanar;
GAGYI PALFFY, Andras, dr.; KICSINDI, Janos, okleveles kohosmernok;
HEINRICH, Jozsef, okleveles banyamernok

The 1963 general meeting of the Hungarian Association for
Mining and Metallurgy. Koh lap 96 no. 6:241-264 Je '63.

1. Chairman, Division of Iron Metallurgy, Hungarian Association for Mining and Metallurgy (for Pitter).
2. Editor-in-Chief, "Kohasziati Lapok" (for Arkos).
3. Secretary, Division of Metallurgy, Hungarian Association for Mining and Metallurgy (for Horvath).
4. Editorial board member, "Kohasziati Lapok" (for Domony).
5. President, Hungarian Association for Mining and Metallurgy (for Levardi).
6. Secretary General, Hungarian Association for Mining and Metallurgy (for Selmecei).
7. Head, Auditing Commission, Hungarian Association for Mining and Metallurgy (for Fekete).
8. Head, Medal Commission, Hungarian Association for Mining and Metallurgy (for Martos).
9. Ozd Metallurgical Works, Ozd (for Macsay).
10. Esztergom Machine Tool Factory, Esztergom (for Kicsindi).